Particle-Associated Polycyclic Aromatic Hydrocarbons and their Dry Deposition Fluxes from a Bus-Station in the Rio de Janeiro Metropolitan Area, Brazil

Mauro C. S. Machado,^a Josiane Loyola,^b Simone L. Quiterio,^{b,c} Gisele O. da Rocha,^{d,e,f} Jailson B. de Andrade^{d,e} and Graciela Arbilla^{*,b}

^aAnalytical Solutions, Rio de Janeiro-RJ, Brazil

^bInstituto de Química, Universidade Federal do Rio de Janeiro, 21949-900 Rio de Janeiro-RJ, Brazil

^cCentro Federal de Educação Tecnológica de Química de Nilópolis, Unidade Maracanã, Rua Senador Furtado, 121, Maracanã, 20270-020 Rio de Janeiro-RJ, Brazil

^dInstituto de Química, Universidade Federal da Bahia, Campus de Ondina, 40170-290 Salvador-BA, Brazil

^eCentro Interdisciplinar de Energia e Ambiente, Universidade Federal da Bahia, 40110-040 Salvador-BA, Brazil

^fInstituto Nacional de Ciência e Tecnologia de Energia e Ambiente, Universidade Federal da Bahia, 40170-290 Salvador-BA, Brazil

Amostras de material particulado total e MP_{10} foram coletadas entre agosto de 2006 e fevereiro de 2007 na estação rodoviária Prefeito José Carlos Lacerda. Foram analisados os 16 HPAs considerados prioritários, emitidos pelos veículos pesados em condições típicas de operação, usando CG/EM. A Σ HPA correspondeu a 0,0018% da massa de MPT e 0,0012% da massa de MP₁₀. O conteúdo total de HPAs foi de 3,57 ng m⁻³ e 2,59 ng m⁻³ para o MPT e o MP₁₀, respectivamente. A contribuição dos compostos carcinogênicos, conforme a USEPA, (B[a]An, B[b]F, B[k]F, B[a] Py, IPy e DB[ah]A) foi 1,86 ng m⁻³ (52% do total de HPAs) e 1,40 ng m⁻³ (54% do total de HPAs), para as amostras de MPT e MP₁₀, respectivamente. As relações características e a análise estatística multivariada deram, em geral, resultados similares aos obtidos para poeira de solos, exceto para a relação Flt/(Flt+Py), 0,34, compatível com emissões de diesel. Os fluxos de deposição seca estão no intervalo 0,022-0,603 e 0,016-0,436 µg m⁻² dia⁻¹ para os compostos contidos no MPT e MP₁₀, respectivamente.

Total particulate matter and PM₁₀ samples were collected from August 2006 to February 2007 in Mayor José Carlos Lacerda bus station. 16 priority PAHs compounds emitted by heavy-duty vehicles on typical conditions of operation were analyzed by using GC/MS. Σ PAH accounted for 0.0018% of the TSP mass and 0.0012% of the PM₁₀ mass. Total PAH contents on the particle phase were 3.57 ng m⁻³ and 2.59 ng m⁻³ for TSP and PM₁₀, respectively. Contributions of carcinogenic USEPA priority PAHs (B[a]An, B[b]F, B[k]F, B[a]Py, IPy and DB[ah]A) were 1.86 ng m⁻³ (52% of total PAHs) and 1.40 ng m⁻³ (54% of total PAHs), for TSP and PM₁₀ samples, respectively. Diagnostic ratios and multivariate statistical analysis were, in general, similar to those reported for soil dust except for the Flt/Flt+Py) ratio, 0.34, compatible with diesel emissions. Dry deposition fluxes were in the range of 0.022-0.603 and 0.016-0.436 µg m⁻² day⁻¹ for compounds in TSP and PM₁₀, respectively.

Keywords: polycyclic aromatic hydrocarbons, TSP, PM₁₀, diagnostic ratios, bus station, dry deposition, diesel emissions

Introduction

The atmosphere is the major transport pathway for the movement of PAHs through the global environment. Once

*e-mail: graciela@iq.ufrj.br

they are released into the atmosphere, PAHs can redistribute between gas and particle phases according to their vapor pressures and atmospheric conditions, and are subjected to removal mechanisms such as oxidative and photocatalytic reactions and wet and dry deposition.^{1,2} The majority of PAHs (70-90%) is sorbed on suspended particles at ambient temperature. Lighter PAHs with 2-3 benzene rings are mostly found in the gas phase while the heavier ones are mainly associated with airborne particles. The carcinogenic larger PAHs (5-7 rings) are associated with particles in the atmosphere. Some of the 3 and 4 rings PAHs are also bound to particles. Also, most of the mutagens in ambient air were proved to be particle-associated. Moreover, PAHs are mostly sorbed on small inhalable particles mainly on airborne particles of submicron diameter which can deposit in the respiratory tract, hence increasing the potential hazardous effects.^{3,4}

PAHs are ubiquitous environmental contaminants. Both the United States Environmental Protection Agency (US EPA) and the European Environmental Agency have mentioned them as priority pollutants. They are originated from both natural and anthropogenic sources. Sources related to energy production are the most important in releasing PAHs. They can be classified as either pyrogenic (mainly from incomplete combustion of organic materials, such as coal, oil, vegetation or fossil fuels) or petrogenic inputs.⁵⁻⁷ PAHs have received attention in air pollution studies because some of these compounds, for instance, benzo[a]pyrene, chrysene, indeno[1,2,3-cd]pyrene, and benzo[b]fluoranthene, have produced carcinogenic, genotoxic and/or mutagenic effects in animals. Accordingly, breathing PAHs may represent a health hazard for humans.8-10

The risk associated with human exposure to atmospheric PAHs is higher in urban environments⁸ where high vehicle density and the dispersion of atmospheric pollutants contribute to the increase of their concentrations. Indeed, in urban areas, exhaust from diesel and gasoline vehicles play an important role on PAH emissions. Otherwise, emissions of PAHs on highways are strongly influenced by heavyduty diesel vehicles.11 Occurrence of PAHs in ambient air is a growing concern due to the continuous nature of exposures and the size of populations at risk, especially in urban, suburban and industrial areas. Therefore, in light of the mutagenicity, carcinogenicity and ubiquity of some PAHs in the atmosphere, the setting of air quality standards and guidelines to limit human exposure should be of primary concern for public health policy. However, difficulties in interpreting heterogeneous experimental data and epidemiological findings9 limit the empirical data needed to establish standards or guidelines.

Several articles were published on the emissions of PAHs from mobile sources.¹²⁻¹⁹ Rogge *et al.*,²⁰ determined the chemical composition of organic fine particulate matter $(PM_{2.5})$ emitted from six non-catalyst vehicles, seven catalyst-equipped automobiles and two heavy duty diesel trucks. More than 100 organic compounds, including

PAHs, were determined in this work. In 1999, Marr et al.21 determined the PAHs concentrations in gasoline and diesel fuel samples collected in northern California and also in particulate matter collected in a tunnel using both filter (PM_{25}) and impactor methods. Emission factors were determined separately for light-duty vehicles and for heavyduty diesel trucks. The authors reported that light-duty vehicles were found to be a significant source of heavier (four- and five-ring) PAHs, whereas heavy-duty diesel engines were the dominant source of three-ring PAHs, such as fluoranthene and pyrene. The obtained results showed no correlation between heavy-duty diesel truck PAH emission factors and PAH concentrations in diesel fuel. In a rather recent study conducted in the centre of Athens (Greece),²² the concentration of PAHs and metals adsorbed to total suspended particulate and finer fractions of airborne particulate matter (PM₁₀ and PM₂₅) were determined. The authors reported that the most common PAHs in $PM_{10.2}$ and $PM_{2.1}$ were pyrene, phenanthrene, acenapthylene and fluoranthene, which are associated with diesel and gasoline exhaust particles. In another study, Turrio-Baldassarri et al.23 compared the emissions of diesel, a blend of diesel and 20% vegetable oil and compressed natural gas engines. The experimental design focused carcinogenic compounds on particulate matter and vapor phase. The results were compared with a previous work²⁴ on the emissions of diesel and biodiesel bus engines. Chrysene followed by benzo[a]anthracene, benzo[b]fluoranthene and benzo[k]fluoranthene were identified as the most abundant compounds.

Most of the data obtained in Brazilian cities refer to total suspended particles.^{6,25-43} Data for inhalable particles and PM25 are rather sparse.40,42,44-47 Some studies were fulfilled in locations highly affected by diesel emissions. Pereira et al.³⁸ determined 16 selected polycyclic aromatic hydrocarbons in total particulate matter, in the Lapa bus station (Salvador, Bahia). Pyrene, fluoranthene and chrysene showed the highest average concentrations. In January 2002, Tavares et al.48 determined the concentrations of vapor phase polycyclic aromatic hydrocarbons at the Londrina Central Bus Station, where only dieselpowered vehicles circulated. The species that presented higher concentration were phenanthrene, fluorene and naphthalene. In 2006, Correa and Arbilla⁴⁹ reported the analysis of the polycyclic aromatic compounds collected in the vapor and aerosol exhaust of a diesel engine. The major compounds were phenanthrene, fluorene and naphthalene. Clearly, PAHs with lower liquid vapor pressures are mainly found in the vapor phase^{38,50} while compounds with vapor pressures lower than 1×10^{-3} Pa are mainly in the aerosol phase.

The objective of this paper is to determine the 16 priority particle-associated (TSP and PM_{10}) PAHs emitted by heavy-duty vehicles, fueled by diesel, on typical conditions of operation. The samplings were performed in a bus station located in the city of Duque de Caxias, in the State of Rio de Janeiro. The studied bus station is poorly ventilated and the intense and frequent movement and parking of buses (which remain with their motors on during passenger accesses) may cause releases of many kinds of hazardous chemicals to the atmosphere. As described bellow, emissions in that location are mainly due to diesel engines. The results are accordingly discussed.

Experimental

Sampling site

The sampling site was the Prefeito José Carlos Lacerda Bus Station, placed in the Centenário Quarter, in the city of Duque de Caxias (22°46'50.44"S, 43°18'28.35"W), State of Rio de Janeiro. The city occupies an area of 468.3 km² and has a population of about 842,890 inhabitants distributed in forty Quarters and four Districts.⁵¹ According to DETRAN, the total number of vehicles in the city is around 160,000 units. Nowadays, 64.7% of vehicles are fueled with gasohol (gasoline with 24% of ethanol), 11.7% with neat ethanol, 12.3% with compressed natural gas (CNG), 2.5% are flexfueled cars (gasohol and ethanol) and 8.8% are heavy-duty diesel vehicles.⁵² The city has about 3,000 buses and 690 micro-buses.

The bus station is located in a residential area with reduced commercial activity and no industries in its surroundings. There is no other activity inside the station except for traffic of buses. The station occupies a 10,621 m² area in one floor and is composed by 8 platforms (each one for four vehicles) for arrivals and departures of urban heavy-duty diesel buses. The confluence of 14 bus lines coming from six other districts of Duque de Caxias and eight other cities of the Metropolitan Region of Rio de Janeiro occurs at the station. Besides the movement of arrival and departure, many vehicles that are waiting for the next departure remain with their motors on. Buses stay about one minute commuting in the station.

No other activities are developed in the bus station and its surroundings. Vehicle count was performed in parallel with particulate matter sampling. The total number of buses is about 180 vehicles a day and light-duty cars represent about 2% of the total flux. No noticeable differences in the vehicle flux were observed during weekdays. Since the only significant emission sources are buses, the location is representative of a diesel impacted environment.

Sampling procedure

The TSP and PM₁₀ samplers were installed beside one of the platforms at a height of 2 m from ground level. TSP and PM₁₀ samplings were performed twice a week using high volume samplers (Sibata, Japan) and borosilicate glass microfiber filters (110 mm diameter, 0.22 mm thickness, Sibata, Japan). Samplings were performed randomly from Monday to Friday. No samplings were performed in the afternoon and at night because the main goal of this work was to identify typical emissions of diesel-powered vehicles. Experiments at night and during weekends were not possible to be performed because of security reasons and availability of samplers. Vehicles were counted in parallel with particulate matter sampling. A flow rate of 500 L min⁻¹ and 6 hour sampling time were adopted for the collection of 66 samples (33 for TSP samples and 33 for PM₁₀ samples) from August 2006 to February 2007 in Prefeito José Carlos Lacerda bus station. After sampling, the filters were wrapped with aluminum foil, placed inside sealed plastic bags and stored at -20 °C up to 14 days until weighing, extraction and analysis. Due to instrumental availability, only TSP and PM₁₀ samples were collected.

Extraction procedure and analysis

Gravimetric determinations were carried out as described in Method IO-3.1, 1999.⁵³

Sixteen PAHs specified on USEPA Method 610⁵⁴ in a mixture, internal standards (naphthalene-D8, acenaphthene-D10, phenanthrene-D10, chrysene-D12, and perylene-D12) and surrogates (2-Fluorobiphenyl and D14-Terphenyl) were obtained from AccuStandard (AccuStandard, USA).

Extraction procedures followed Method 3550b⁵⁵ as described by Quiterio *et al.*⁴² Briefly, surrogate mix was added to the filter samples before extraction procedure, then those filters were sonicated four times with 50 mL dichloromethane (DCM) for 15 min and then with 100 mL for 20 min.⁵⁶ The organic extract was filtered and concentrated to 2 mL on a rotatory evaporator. Surrogate recoveries ranged from 48 to 72%. Afterwards, PAH concentrations were corrected according to the efficiency of recovery during extraction.

Prior to injection, 1 μ g of internal standards mix was added to the extracts. All samples were analyzed using a Gas Chromatograph (GC Agilent Model 6890, USA) hyphenised to a Mass Spectrometry selective detector (Agilent Model 5973, USA), equipped with a HP-5 (30 m × 0.25 mm dia. × 0.25 μ m film thickness, Agilent, USA) GC column. 1 µL samples were injected on splitless mode at an injector temperature of 300 °C. The carrier gas was helium at a constant flow of 1.2 mL min⁻¹. Oven temperature was programmed as follows: 50 °C (2 min) \rightarrow 300 °C (12 °C min⁻¹) \rightarrow 300 °C (10 min). Quantitative analyses were performed using a single ion monitoring (SIM) method under 70 eV of electron energy. Chromatographic peaks were identified and quantified, based on their GC-MS acquired retention times and mass spectra of the authentic standards used as reference: naphthalene (Na)-128, acenaphthylene (Acy)-152, acenaphthene (Ace)-153, fluorene (Flu)-166, phenanthrene (Phe)-178, anthracene (An)-178, fluoranthene (Flt)-202, pyrene (Py)-202, benzo[a]anthracene (B[a]An)-228, chrysene (Chry)-228, benzo[b]fluoranthene (B[b]F)-252, benzo[k]fluoranthene (B[k]F)-252, benzo[a]pyrene (B[a] Py)-252, indene[1,2,3-c,d]pyrene (IPy)-276, dibenz[a,h] anthracene (DB[ah]A)-278, and benzo[ghi]perylene (B[ghi]P)-276. Field blanks, which accompanied samples from the collecting sites, were used in order to avoid background contamination of results.

Quantification limit (QL) and detection limit (DL) were determined as 10 ng and 8 ng, respectively. The QL was the first point (the point of lowest concentration) from each linear calibration curve. DL was determined from the standard deviation from the responses of seven replicates of the lowest point in the calibration curve, multiplied by the correspondent t-factor for 95% confidence interval. Considering the total sampled air volume (1600 m³), the QL and DL from this method were estimated to be 0.006 ng m⁻³ and 0.005 ng m⁻³, respectively.

Statistical analysis

Experimental data were analysed using STATISTICA 6.0 (Statsoft, USA) software pack. Principal Component Analysis (PCA) and Cluster Analysis (CA), using Euclidian distances and Ward's Method, were also performed, as classification and ordination methods.

Results and Discussion

PAH levels in TSP and PM_{10} samples

TSP levels were in the range of 39 to 52.4 μ g m⁻³ and the arithmetic mean was 200 μ g m⁻³. Similarly, values for PM₁₀ ranged from 38 to 434 μ g m⁻³, with a mean value of 210 μ g m⁻³.

The concentrations of the studied PAH samples are shown in Table 1. The average values (in ng m⁻³) are mean values for the 33 samples (33 TSP and 33 PM_{10} samples). Individual concentrations were between the detection limit for acenaphtylene, acenaphthene, anthracene and fluorene and 2.27 and 1.74 ng m⁻³ for benzo[ghi] perylene (B[ghi]P) for TSP and PM₁₀, respectively. The concentrations of naphthalene (Na), acenaphtylene (Acy), acenaphthene (Ace), fluorene (Flu) are not reported because samplings were performed for rather long periods (6 h) in unfavorable temperature conditions. The platform area is not well ventilated and samples were collected in spring and summer when ambient temperatures are very high in Brazil. The obtained concentrations values are probably underestimated due to volatilization of the low molecular weight compounds. Long sampling periods may alter the chemical composition of samples because of volatilization, chemical and photochemical transformations leading to underestimated PAH concentrations. A comparative study of PAH concentrations determined in PM₁₀ and GF/PUF (glass fiber filters and a polyurethane foam filter) samples collected at two urban stations in the city of Brno, Czech Republic,⁵⁷ showed that the GF/PUF method gave an amount of low molecular weight U.S.EPA priority PAHs (Na, Acy, Ace, Flu, Phe, An, Flt, Py) several times higher than the high volume sampling of the PM₁₀ fraction. Ciganek et al.⁵⁷ concluded that sampling using the PM₁₀ method could lead to an underestimation of overall PAH concentrations and, particularly, of low molecular weight compounds.

 Σ PAH accounted for 0.0018% of the TSP mass and 0.0012% of the PM₁₀ mass. The total PAHs contents on

Table 1. PAH concentration (ng m⁻³) and PAH particle content of PM collected in this study (units in µg g⁻¹)

	Phe	Flt	Ру	B[a]An	Chry	B[b]F	B[k]F	B[a]Py	Ipy	DB[ah]A	B[ghi]P	Total
PAH concentration / (ng m ⁻³)												
TSP	0.116	0.174	0.288	0.166	0.262	0.557	0.131	0.363	0.524	0.117	0.866	3.564
PM ₁₀	0.087	0.119	0.203	0.132	0.191	0.424	0.107	0.268	0.365	0.118	0.611	2.625
PAH content / ($\mu g g^{-1}$)												
TSP	0.58	0.87	1.44	0.83	1.31	2.78	0.65	1.81	2.61	0.58	4.32	17.78
PM ₁₀	0.41	0.57	0.97	0.63	0.91	2.02	0.51	1.28	1.74	0.56	2.91	12.49

the particle phase were 17.8 μ g g⁻¹ and 12.5 μ g g⁻¹ for TSP and PM₁₀, respectively. Individual PAH content on both particle phase fractions are described on Table 1. Contribution of carcinogenic USEPA priority PAHs (B[a]An, B[b]F, B[k]F, B[a]Py, IPy and DB[ah]A) were 9.27 μ g g⁻¹ (52% of total PAHs) and 6.73 μ g g⁻¹ (54% of total PAHs), respectively. Carcinogenic PAHs contributions show similar percentual values thus evidencing serious health concerns.

Unfortunately, there are few studies reporting PAH concentrations in Brazilian areas emitted from heavyduty diesel vehicles.38,48 Tavares Jr. et al.48 determined the concentration of vapor phase PAHs at the central bus station of Londrina (Brazil) where only diesel-powered vehicles circulate. PAHs were collected using XAD-2 resin cartridges and analyzed by gas chromatography with both flame ionization and mass spectrometer detectors. Ten PAH were found (naphthalene, acenaphthylene, acenaphthene, fluorene, phenanthrene, anthracene, fluoranthene, pyrene, benzo(a)anthracene and chrysene) and the most abundant compounds were phenantrene, fluorene and naphthalene. Pereira et al.³⁸ used a TSP high volume sampler with quartz fiber filters. Samples were collected at the Lapa bus station located in a downtown area of Salvador, Bahia (Brazil). Chrysene presented the highest average concentration but relative high concentrations were obtained for fluorantene, pyrene, benzo[a]anthracene, benzo[b]fluoranthene, benzo[k] fluoranthene, benzo[a]pyrene, indene[1,2,3-cd]pyrene and benzo[ghi]perylene as well. Our results are in quite good agreement with these data.

Diagnostic ratios

PAH concentrations and molecular ratios have been used to distinguish emissions and indicate the impact of different sources of airborne compounds.⁵⁸⁻⁶²

Diagnostic ratios between PAHs are presented in Table $2^{12,20,38,63-68}$ both for TSP and PM₁₀ samples. Ratios are very similar for PM₁₀ and TSP samples. Values obtained in PM₁₀ samples are indicated in italics.

In 2004, Manoli et al.12 published the PAHs profiles for several urban, industrial and geological sources, in particular diesel fueled buses, paved road dust and soil dust absorbed on PM_{10} samples. In general, diagnostic ratios for ambient urban samples were similar to those obtained for paved road and soil dust. Ratios obtained from the exhaust of diesel vehicles are rather different suggesting that the PM₁₀ fraction has an important contribution of mechanical sources and ressuspension of dust. In Table 2, typical ratios for diesel vehicles, light cars, urban TSP and PM₁₀ samples, paved road dust, soil dust are listed for comparison. Urban samples were those determined by Mantis et al.63 in an Aristotelous street, in downtown Athens, with dense vehicular traffic and commercial activities, and by Manoli et al.¹² in the city center of Thessalonike, Greece, an area with high commercial activities and traffic density. It must be noted that the reported ratios are, in many cases, spread in a wide range of values and there is not a clear difference between diesel and gasoline emissions.

The sum of concentrations of the nine combustion PAHs (CPAH) (see Table 2) divided by the total concentration of PAHs gave the values 0.93 and 0.91 for TSP and PM₁₀

Table 2. Diagnostic PAH ratios of samples obtained at Mayor José Carlos Lacerda Bus Station

	This study TSP	This study PM ₁₀	Diesel vehicles	Gasoline vehicles	Urban Samples	Soil dust	Paved road dust
СРАН/∑РАН	0.93 ± 0.03	0.91 ± 0.17	0.77; ¹² 0.88 ⁶³	0.73;12 0.6068	$\begin{array}{c} 0.95;^{12} \ 1.18;^{33} \\ 0.92^{63} \end{array}$	0.89;12 0.8312	$\begin{array}{c} 0.82;^{12} \ 0.86;^{12} \\ 0.87^{12} \end{array}$
B[ghi]Pe/B[a]Py	2.44 ± 0.80	2.27 ± 0.88	$\begin{array}{c} 0.11;^{12} \ 3.69;^{65} \\ 1.2\text{-}2.2^{64} \end{array}$	$3.05;^{12} 1.72;^{12} \\ 2.5-3.3^{64}$	$2.00;^{12} 1.26;^{33} \\ 4.36^{63}$	$1.18;^{12} 0.25;^{12} \\ 0.86^{12}$	$\begin{array}{c} 2.70;^{12} \ 1.09;^{12} \\ 3.67;^{12} \ 0.91^{20} \end{array}$
IPy/(Ipy + B[ghi]Pe)	0.37 ± 0.05	0.37 ± 0.05	0.96; ¹² 0.35- 0.70 ⁶⁴	0.26;12 0.5112	$\begin{array}{c} 0.38;^{12} \ 0.38;^{33} \\ 0.32^{63} \end{array}$	0.61; ¹² 0.91 ¹²	$\begin{array}{c} 0.51;^{12} \ 0.57;^{12} \\ 0.52^{12} \end{array}$
B[a]An/(B[a]An + Chry)	0.39 ± 0.03	0.40 ± 0.08	0.73;12 0.3763	0.76;12 0.5812	$\begin{array}{c} 0.43;^{12} \ 0.20;^{33} \\ 0.40^{63} \end{array}$	0.29;12 0.2012	$\begin{array}{c} 0.67;^{12} \ 0.42;^{12} \\ 0.32;^{12} \ 0.50^{65,66} \end{array}$
Flt/(Flt + Py)	0.37 ± 0.03	0.36 ± 0.07	$\begin{array}{c} 0.38;^{12} \ 0.46;^{63} \\ 0.36^{20} \end{array}$	$\begin{array}{c} 0.14;^{12} \ 0.17;^{12} \ 0.47;^{67} \\ 0.41;^{65,66} \ 0.40^{64} \end{array}$	$\begin{array}{c} 0.37;^{12} \ 0.34;^{33} \\ 0.39^{63} \end{array}$	0.52;12 0.6112	$\begin{array}{c} 0.42;^{12} \ 0.52;^{12} \\ 0.42^{64} \end{array}$
B[a]Py/(B[a]Py + Chry)	0.56 ± 0.12	0.55 ± 0.15	$0.38-0.65;^{65,66}$ 0.46^{68}	na	0.1833	na	na

CPAH: Flt +Py+B[a]An+Chry+B[b]F+B[k]F+B[a]Py+B[ghi]Pe+IPy.

¹²Manoli *et al.*; ³³Pereira Netto *et al.*; ⁶³Mantis *et al.*; ²⁰Rogge *et al.*; ⁶⁴Rogge *et al.*; ⁶⁵Kavouras *et al.*; ⁶⁶Kavouras *et al.*; ⁶⁷Sienra *et al.*; ⁶⁸Oda *et al.*; na: not available.

samples, respectively. This value is in the same range of previous reported ratios for downtown areas.^{12,63} As shown in Table 2, these values are also similar to those obtained for soil and road dust samples.

The ratios B[ghi]Pe/B[a]Py, calculated as 2.44 and 2.27 for TSP and PM₁₀, respectively, are compatible with values reported for road dust and some urban areas.⁶³ Values obtained directly from buses exhausts are significantly smaller, ranging from 0.11^{12} to 1.2- $2.2.^{63}$

Considering the carcinogenic compounds, the ratios, B[a]An/(B[a]An + Chry) and B[a]Py/(B[a]Py + Chry) were calculated as 0.38-0.39 and 0.56-0.55, respectively, values also associated to urban areas with high density traffic and road dust.^{12,63,65,66} The value determined from the direct exhaust of diesel buses is 0.73.¹²

The ratio Flt/(Flt+Py) was calculated as 0.37 and 0.36 for TSP an PM₁₀, respectively. These values are in good agreement with the ratio obtained for diesel buses,¹² diesel trucks²⁰ and urban areas^{12,63} and they are also similar to the value obtained by Quiterio *et al.*⁴² in an area with a high contribution of diesel buses and trucks. These data may be compared with those determined by Pereira *et al.*³⁸ at the Lapa bus station in Salvador for TSP samples. The diagnostic ratios displayed in Table 2 were calculated using the mean concentrations reported by the authors. The value for the CPAH/ Σ PAH ratio is similar to that reported in this work. The other ratios are, in general, similar to those reported for soil dust¹² except for the Flt/(Flt+Py) ratio, 0.34, compatible with diesel emissions.

The benzo[a]pyrene equivalent carcinogenic power (BaPE)^{69,70} was also estimated. This index tries to parameterize the health risk for humans related to ambient PAH exposition and is calculated by multiplying the concentrations of each carcinogenic congener, as follows:

 $B[a]An \times 0.06 + {B[b]F + B[k]F} \times 0.07 + B[a]Py + DB[ah]An \times 0.6 + IPy \times 0.08.$

The calculated values are 0.17 and 0.41 ng m⁻³ for TSP and PM₁₀ samples, respectively. These values do not represent a significant cancer risk for passengers and workers in the bus station area.

This is, to our knowledge, the first reported diagnostic ratio for a diesel fleet in Brazil and may be considered as a typical value for emissions of high-duty vehicles in the country. Anyway it is clear that a further study in the PM_{2.5} fraction is highly desirable in order to assess both the inhalable fraction and the fraction due predominately to combustion processes.

Statistical analysis

In order to obtain an insight on the main correlations among PAHs, cluster analysis (CA) and factorial analysis (FA) were applied for each set of data.

Table 3. Results of PCA for the correlation matrix $[33 \times 11]$ obtained with TSP and PM₁₀ data at the bus station

	T: [33 :	PM_{10} [33 × 11]			
PC	1	2	1	2	
Eigenvalues	7.4	1.6	7.5	1.5	
% total variance	66.8	14.1	68.0	13.8	
Loading					
Phe	0.7	0.4	0.7	0.4	
Flt	0.9	0.2	0.8	0.3	
Ру	0.9	0.2	0.8	0.3	
B[a]An	0.2	0.9	0.2	0.9	
Chry	0.2	1.0	0.2	0.9	
B[b]F	0.8	0.2	0.9	0.2	
B[k]F	0.9	0.2	0.8	0.4	
B[a]Py	0.9	0.3	0.8	0.4	
IPy	0.8	0.3	0.8	0.5	
DB[ah]An	0.9	0.0	0.9	-0.1	
B[ghi]Pe	0.7	0.4	0.6	0.6	

Factor loadings > 0.5 are in bold.

For TSP samples (Figure 1, Table 3), the correlation matrix $[33 \times 11]$ gave two main clusters and two significant principal components (PCs) which explain 81% of the total variance. The elements Chry and B[a]An are together in PC2 and in the same cluster. The other elements are in PC1 and also in the other cluster. The correlation matrix $[33 \times 11]$ gave the same cluster and PCs for PM₁₀ samples

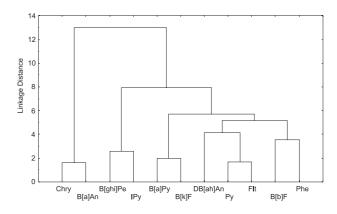


Figure 1. Dendogram of the cluster analysis of PAHs at the bus station for TSP samples.

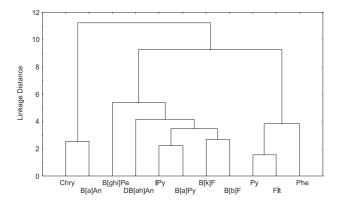


Figure 2. Dendogram of the cluster analysis of PAHs at the bus station for PM_{10} samples.

(Figure 2, Table 3) than for total particulate matter. The only noticeable difference is that B[ghi]Pe is associated both to PC1 and PC2. Also, two subgroups may be identified within the second cluster according to their liquid vapor pressures.

The Flt/Py correlation coefficient was 0.96, in close agreement with that obtained by Pereira *et al.*³⁸ It is noteworthy that these were the only compounds that gave a diagnostic ratio compatible with direct diesel emissions.

Particle-associated PAHs dry deposition fluxes

Dry atmospheric deposition fluxes (F_d) were calculated by multiplying the geometric mean of each PAH concentration (C_i) in the particulate matter times the PAH dry settling velocity (V_d) as follow:

$$\mathbf{F}_{d} = \mathbf{C}_{i} \times \mathbf{V}_{d} \tag{1}$$

Deposition velocity, $V_{d_{,}}$ may vary 1-2 orders of magnitude^{8,71-73} due either to particle size, climatic or physical conditions in the atmosphere Even considering those uncertainties, the particulate matter dry deposition flux calculations help to better understand PAHs removal

mechanisms. Indeed, some authors, such as Vardar *et al.*⁷¹ and Chang *et al.*,⁸ have modeled V_d and they have found results comparable to those from Sheu *et al.*⁷² In this study PAH particle dry depositions were calculated using V_d as stated on the latter cited study as shown in Table 4.

Dry deposition flux for the total PAHs (F_d of Σ PAH) showed the highest value (0.603 µg m⁻² day⁻¹) for TSP followed by PM₁₀ (0.436 µg m⁻² day⁻¹). Individual deposition fluxes are found in Table 4. In most sites B[ghi]P was the PAH with the highest deposition flux followed by IPy.

The apparent dry deposition fluxes from Table 4 were calculated taking into account both the geometric average concentration level of an individual PAH sorbed on particulate matter and the dry deposition velocity but it should also have been taken into account PAH reactivity which can be an important parameter in interpreting the dataset. Nilsen⁷⁴ has developed a reactivity scale that groups PAHs into five classes of reactivity (from Class I-the most reactive group-to Class V-the least reactive one) toward nitrating species generating then either nitro-PAHs, oxy-PAHs or quinones (these are PAH derivatives much more carcinogenic and/or mutagenic than the originating species and then of greater concern in health issues). If a PAH is more reactive than other, it is more easily modified (removed) by a photochemical reaction and it would be not found at high levels in PM and then dry or wet deposition mechanisms would not be its main atmospheric fate. On the other hand, the less reactive the PAH the more probable is to sink by either dry or wet depositions, depending on its vapor pressure and water solubility (being, therefore, probably promoted to any terrestrial and/or aquatic systems). This should be considered when analyzing some isomer pairs such as Phe and An, Flt and Py, B[b]F and B[k]F, and IPy and DB[ah]A. Flt belongs to Class V and Py is Class III then it is reasonable to accept that Py, being more reactive than Flt, would be more readily modified to any of its possible nitro-derivatives and less Py would be available to be dry deposited.75 The same occurs with IPy (Class V)

Bus Station, Duque de Caxias, RJ	Phe	Flt	Ру	B[a]An	Chry	B[b]F	B[k]F	B[a]Py	IPy	DB[ah]A	B[ghi]Pe
Dry settling velocity / (V_d) (cm s ⁻¹)*	0.23	0.41	0.2	0.35	0.54	0.55	0.62	0.71	0.89	0.76	0.97
TSP											
Geometric mean / (ng m ⁻³)	0.11	0.15	0.25	0.15	0.24	0.48	0.12	0.31	0.41	0.10	0.72
Dry deposition flux (F_d) / ($\mu g m^{-2} day^{-1}$)	0.022	0.053	0.043	0.045	0.112	0.228	0.166	0.190	0.315	0.066	0.603
PM ₁₀											
Geometric mean / (ng m ⁻³)	0.080	0.13	0.18	0.12	0.17	0.37	0.097	0.24	0.30	0.11	0.52
Dry deposition flux (F_d) / ($\mu g m^{-2} day^{-1}$)	0.016	0.046	0.031	0.036	0.079	0.176	0.052	0.147	0.231	0.072	0.436

Table 4. Estimates of particle PAH dry deposition fluxes (F_d)

*According to Sheu et al.,⁷² 1996.

and DB[ah]A (Class IV), the latter apparently possessing lower dry deposition fluxes. In the case of B[b]F and B[k] F pair, both belong to the same reactivity class (Class V) and their high fluxes derive from both high concentration levels and deposition velocities. Summing up, if dry deposition were the main removal mechanism for Phe, Flt, IPy, B[b]F, B[k]F and in minor extension, for DB[ah] A (and the other Class IV, Chry), those particle airborne once deposited might be resuspended by any mechanical/ physical perturbation being then able to be enriched of some PAH freshly generated (vapor-PAH converted to particle or by any physical interaction as accumulation, coagulation, etc acting in both freshly and aged particulate PAH) and be again dry deposited. This cyclic path of the PAHs could occur continuously and the particulate matter be aged in relation to some less reactive PAH.

Conclusions

In this work the atmospheric concentrations of priority PAHs associated to TSP and PM_{10} samples were determined. Individual concentrations were between the detection limit (acenaphtylene, acenaphthene, anthracene and fluorene) and 2.27 and 1.74 ng m⁻³ for benzo[ghi] perylene (B[ghi]P) for TSP and PM₁₀, respectively.

ΣPAH accounted for 0.0018% of the TSP mass and 0.0012% of the PM₁₀ mass. The total PAH contents on the particle phase were 17.8 μg g⁻¹ and 12.5 μg g⁻¹ for TSP and PM₁₀, respectively. Contributions of carcinogenic USEPA priority PAHs (B[a]An, B[b]F, B[k]F, B[a]Py, IPy and DB[ah]A) were 9.27 μg g⁻¹ (52% of total PAHs) and 6.73 μg g⁻¹ (54% of total PAHs), respectively, then, a serious health concern.

Diagnostic ratios were, in general, quite different from those determined in the exhaust of diesel vehicles suggesting an important contribution of mechanical sources and ressuspension of dust. A further study in the respirable fraction is highly desirable in order to assess the contribution of PAHs due predominantly to combustion processes.

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References

 Mastral, A. M.; Lopez, J. M.; Callen, M. S.; Garcia, T.; Murillo, R.; Navarro, M. V.; *Sci. Total Environ.* **2003**, *307*, 111.

- 2. Bodnar, E.; Hlavay, J.; Microchem. J. 2005, 79, 213.
- 3. Nielsen, T.; Atmos. Environ. 1996, 30, 3481.
- 4. Bae, S. Y.; Yi S. M.; Kim Y. P.; Atmos. Environ. 2002, 36, 5491.
- Meudec, A.; Dussauze, J.; Deslandes, E.; Poupart, N.; Chemosphere 2006, 65, 474.
- Vasconcellos, P. C.; Zacarias, D.; Pires, M. A. F.; Pool, C. S.; Carvalho, L. R. F.; *Atmos. Environ.* 2003, *37*, 3009.
- Albinet, A.; Leoz-Garziandia, E.; Budzinski, H.; Villenave, E.; J. Chromatogr., A 2006, 1121, 106.
- Chang, K. F.; Fang, G. C.; Lu, C.; Bai, H.; *Aerosol Air Qual. Res.* 2003, *3*, 41.
- 9. Chen, S. C.; Liao, C. M.; Sci. Total Environ. 2006, 366, 112.
- Dabestani, R.; Ivanov, I. N.; *Photochem. Photobiol. Sci.* 1999, 70, 10.
- Shah, S. D.; Ogunyoku, T. A.; Miller, J. W.; Cocker, D. R.; Environ. Sci. Technol. 2005, 39, 5276.
- Manoli, E.; Kouras, A.; Samara, C.; *Chemosphere* 2004, 56, 867.
- Fromme, H.; Oddoy, A.; Piloty, M.; Krause, M.; Lahrz, T.; Sci. Total Environ. 1998, 217, 165.
- Kendall, M.; Duarte, A.; Rocha-Santos, T.; Hamilton, R.; Williams, I.; J. Environ. Monit. 2002, 4, 890.
- Hien, T. T.; Nam, P. P.; Yasuhiro, S.; Takayuki, K.; Norimichi, T.; Hiroshi, B.; *Sci. Total Environ.* 2007, *382*, 70.
- Abrantes, R.; Assunção, J. V.; Pesquero, C. R.; *Atmos. Environ.* 2004, *38*, 1631.
- Devos, O.; Combet, E.; Tassel, P.; Paturel, L.; *Polycyclic Aromat. Compd.* 2006, 26, 69.
- Durán, A.; de Lucas, A.; Carmona, M.; Ballesteros, R; Chemosphere 2001, 44, 921.
- Yang, H. H.; Chien, S. M.; Chao, M.-R.; Lin, C. C.; J. Hazard. Mater. 2005, 125, 154.
- Rogge, W. F.; Hildemann, L. M.; Mazurek, M. A.; Cass, G. R.; Simoneit, B. R. T.; *Environ. Sci. Technol.* **1993**, *27*, 636.
- Marr L.C.; Kirchstter W.; Harley R. A.; Miguel A. H.; Herings S. V.; Hammonds S. K.; *Environ. Sci. Technol.* 1999, *33*, 3091.
- Valavanidis, A.; Fiotakis, K.; Vlahogianni, T.; Bakeas, E. B.; Triantafillaki, S.; Paraskevopoulou, V.; Dassenakis, M.; *Chemosphere* 2006, 65, 760.
- Turrio-Baldassarri, L.; Battistelli, C.L.; Conti, L.; Crebelli, R.; De Berardis, B; Iamiceli, A.L.; Gambino, M.; Iannaccone, S.; *Sci. Total Environ.* 2006, *355*, 64.
- Turrio-Baldassarri, L.; Battistelli, C.L.; Conti, L.; Crebelli, R.; De Berardis, B; Iamiceli, A.L.; Gambino, M.; Iannaccone, S.; *Sci. Total Environ.* 2004, *327*, 147.
- Vasconcellos, P. C.; Zacarias, D.; Carvalho, L. R. F.; Pires, M. A. F.; Pool, C.; *Atmos. Environ.* 2003, *37*, 3009.
- Azevedo, D. A.; Moreira, L. S.; Siqueira, D. S.; *Atmos. Environ.* 1999, *33*, 4987.
- Oliveira, A. S.; Fernandes, M. B.; Moreira, J. C.; Ferreira, L. F. V.; *J. Braz. Chem. Soc.* 2002, *13*, 245.

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- Fernandes, M. B.; Brickus, L. S. R.; Moreira, J. C.; Cardoso, J. N.; *Chemosphere* **2002**, *47*, 417.
- Ré-Poppi, N.; Santiago-Silva, M.; *Atmos. Environ.* 2005, *39*, 2839.
- Ströher, G. L.; Ré-Poppi, N.; Raposo Jr., J. L.; Souza, J. B. G.; Microchem. J. 2007, 86, 112.
- Pereira Netto, A. D.; Cunha, I. F.; Muniz, F. C.; Rego, E. C. P.; Bull. Environ. Contam. Toxicol. 2001, 67, 271.
- Pereira Netto, A. D.; Muniz, F. C.; Rego, E. C. P.; *Bull. Environ. Contam. Toxicol.* **2002**, *68*, 831.
- Pereira Netto, A. D.; Barreto, R. P.; Moreira, J. C.; Arbilla, G.; Bull. Environ. Contam. Toxicol. 2002, 69, 173.
- Pereira Netto, A. D.; Sisinno, C. L.; Moreira, J. C.; Arbilla, G.; Dufrayer M. C., *Bull. Environ. Contam. Toxicol.* 2002, 68, 148.
- Pereira Netto, A. D.; Krauss, T. M.; Cunha, I. F.; Rego, E.C.P.; Water, Air, Soil Pollut. 2006, 176, 57.
- Lima, F. M.; *MSc Dissertation*, Universidade Federal Fluminense, Brazil, 2006.
- Azevedo, D. A.; Dos Santos, C. Y. M.; Aquino Netto, F. R.; Atmos. Environ. 2002, 36, 2383.
- Pereira, P. A. P.; de Andrade, J. B.; Miguel, A. H.; *J. Environ. Monit.* 2002, *4*, 558.
- Miguel, A. H.; Pereira, P. A. P.; *Aerosol Sci. Technol.* 1989, 10, 292.
- Martinis, B.; Okamoto, R. A. B.; Kado, N. Y.; Gundel, L.; Carvalho, L. R. F.; *Atmos. Environ.* **2002**, *36*, 307.
- Santos, C. Y. M.; Azevedo, D. A.; Aquino Neto, F. R.; *Atmos. Environ.* 2002, *36*, 3009.
- Quiterio, S. L.; Arbilla, G.; Bauerfeldt, G. F.; Moreira, J. C.; Water, Air, Soil Pollut. 2007, 179, 79.
- 43. Miguel, A. H.; de Andrade, J. B.; Hering, S. V.; *Int. J. Environ. Anal. Chem.* **1986**, *26*, 265.
- Bourotte, C.; Forti, M. C.; Taniguchi, S.; Bícego, C.; Lotufo, P. A.; *Atmos. Environ.* 2005, *39*, 3799.
- Dallarosa, J. B.; Mônego, J. G.; Teixeira, E. C.; Stefens, J. L.; Wiegand, F.; *Atmos. Environ.* **2005**, *39*, 1609.
- Dallarosa, J. B.; Teixeira, E. C.; Meira L.; Wiegand, F.; *Atmos. Environ.* 2008, 89, 76.
- Dallarosa, J. B.; Teixeira, E. C.; Pires, M.; Fachel, J.; *Atmos. Environ.* 2005, *39*, 6587.
- Tavares Jr, T.; Pinto, J. P.; Souza, A. L.; Scarmínio, I. S.; Solci, M. C.; *Atmos. Environ.* **2004**, *38*, 5039.
- 49. Corrêa, S. M.; Arbilla, G.; Atmos. Environ. 2006, 40, 6821.
- Wingfors, H.; Sjodin, A.; Haglund, P.; Brorstrom-Lundén, E.; Atmos. Environ. 2001, 35, 6361.
- http://www.duquedecaxias.rj.gov.br/web/cgi/cgilua.exe/sys/ start.htm, accessed in April 2007.

- 52. http://www.detran.rj.gov.br, accessed in July 2007.
- http://www.epa.gov/ttnamti1/files/ambient/inorganic/mthd-3-1. pdf, accessed in July 2006
- http://www.epa.gov/Region9/qa/pdfs/40cfr136_03.pdf, accessed in July 2006
- http://www.epa.gov/epaoswer/hazwaste/test/pdfs/3550c.pdf, accessed in July 2006
- http://www.epa.gov/epaoswer/hazwaste/test/pdfs/8270c.pdf , accessed in July 2006
- Ciganek, M.; Neca, J.; Adamec, V.; Janosek, J.; Machala, M.; Sci. Total Environ. 2004, 334-335, 141.
- Guo, H.; Lee, S. C.; Ho, K. F.; Wang, X. M.; Zou, S. C.; *Atmos. Environ.* 2003, *37*, 5307.
- Venkataraman, C.; Lyons, J. M.; Friedlander, S. K.; *Environ. Sci. Technol.* **1994**, *28*, 555.
- Cotham, W. E.; Bidleman, T. F.; *Environ. Sci. Technol.* 1995, 29, 2782.
- Harrison, R. M.; Smith, D. J. T.; Luhana, L.; *Environ. Sci. Technol.* **1996**, *30*, 825.
- 62. Alves, C.; Pio, C.; Duarte, A.; Atmos. Environ. 2001, 35, 5485.
- Mantis, J.; Chaloulakou, A.; Samara, C.; *Chemosphere* 2005, 59, 593.
- Rogge, W. F.; Hildemann, L. M.; Mazurek, M. A.; Cass, G. R.; Simoneit, B. R. T.; *Environ. Sci. Technol.* **1993**, *27*, 1892.
- Kavouras, I. G.; Lawrence, J.; Koutrakis, P.; Stephanou, E. G.; Oyola, P.; *Atmos. Environ.* **1999**, *33*, 4977.
- Kavouras, I. G.; Koutrakis, P.; Tsapakis, M.; Lagoudaki, E.; Stephanou, E. G.; Von Baer, D.; Oyola, P.; *Environ. Sci. Technol.* 2001, *35*, 2288.
- Sienra, M. R.; Rosazza, N. G.; Préndez, M.; *Atmos. Res.* 2005, 75, 267.
- Oda, J.; Nomura, S.; Yasuhara, A.; Shibamoto, T.; *Atmos. Environ.* 2001, *35*, 4819.
- 69. Alves, C.; An. Acad. Bras. Ciênc. 2008, 80, 21.
- Marino, F; Cecinato, A; Siskos, P.A.; *Chemosphere* 2000, 40, 533.
- Vardar, N.; Odabasi, M.; Holsen, T. M.; J. Environ. Eng. 2002, 128, 269.
- Sheu, H.-L.; Lee, W. J.; Su, C. C.; Chao, H. R.; Fan, Y. C.; J. Environ. Eng. 1996, 122, 1101.
- Fang, G. C.; Chang, K. F.; Lu, C.; Bai, H.; *Chemosphere* 2004, 55, 787.
- 74. Nielsen, T.; Environ. Sci. Technol. 1984, 18, 157.
- 75. Tasdemir, Y.; Esen, F.; Atmos. Environ. 2007, 41, 1288.

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